

## Research Article

# An asymmetric synthesis of L-[3-<sup>13</sup>C]alanine

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## Summary

L-[3-<sup>13</sup>C]Alanine was synthesized from [<sup>13</sup>C]methyl iodide by using Dellaria's oxazinone, prepared from phenyl[2-<sup>13</sup>C]bromoacetate and (*S*)-2-phenylglycinol, as a chiral glycine equivalent. Alkylation of the oxazinone with [<sup>13</sup>C]methyl iodide was achieved with high diastereoselectivity. Hydrolysis and removal of the chiral auxiliary of the alkylated oxazinone gave L-[3-<sup>13</sup>C]alanine. Copyright © 2003 John Wiley & Sons, Ltd.

**Key Words:** L-[3-<sup>13</sup>C]alanine; [<sup>13</sup>C]methyl iodide; chiral glycine equivalent; oxazinone

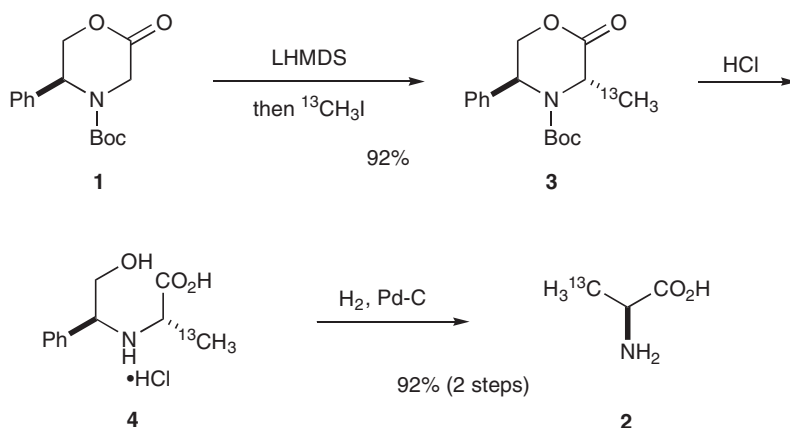
## Introduction

Stable isotope-labelled amino acids are useful for the diagnosis of disease, the study of the biosynthesis of natural products and other biological studies.<sup>1–3</sup> We described previously asymmetric syntheses of L-[3-<sup>13</sup>C]phenylalanine, L-[3-<sup>13</sup>C]tyrosine<sup>4</sup> and L-[2-<sup>13</sup>C]aspartic acid<sup>5</sup> from Dellaria's oxazinone **1** or its <sup>13</sup>C-labelled form as a chiral glycine equivalent. This method is also applicable to the synthesis of various amino acids with labelling at other positions. In this paper, we describe an asymmetric synthesis of L-[3-<sup>13</sup>C]alanine (**2**) (For a recent synthesis of optically active <sup>13</sup>C-labelled alanine from a chiral glycine equivalent, see ref. 7 and 8.).

## Results and discussion

Dellaria's oxazinone **1** was prepared from (*S*)-2-phenylglycinol and bromoacetyl bromide by modifying the method of Dellaria, described in a previous report.<sup>5</sup> Diastereoselective alkylation of **1** was performed by treatment of **1** with sodium bis(trimethylsilyl)amide (NaHMDS) in THF-DME and addition of [<sup>13</sup>C]methyl iodide to the resulting enolate of **1** to give the alkylated

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**Scheme 1.**

oxazinone **3** in 92% yield with high diastereoselectivity (Scheme 1). Hydrolysis of the oxazinone ring and removal of the Boc group of **3** were simultaneously carried out by heating at 80°C in 6 M HCl to afford *N*-alkylated L-[3-<sup>13</sup>C]alanine **4**, which was used in the next step without purification. Finally, hydrogenolysis of **4** gave L-[3-<sup>13</sup>C]alanine (**2**) in 92% yield from **3**. The overall yield from [<sup>13</sup>C]methyl iodide to L-[3-<sup>13</sup>C]alanine (**2**) was 84%.

The <sup>13</sup>C-NMR spectrum of **2** showed the enriched signal at 19.0 ppm, assigned to the 3-position of alanine. The <sup>1</sup>H-NMR spectrum showed a strong signal at 1.49 ppm as a double-doublet ( $J_{C-H} = 129.9$  Hz and  $J_{H-H} = 7.2$  Hz), assigned to the 3-position of **2**, and a weak signal as a doublet ( $J_{H-H} = 7.2$  Hz), assigned to the same position of residual unlabelled alanine. The enrichment ratio was calculated at 99 atom% <sup>13</sup>C or above, based on the integration ratio of these signals. The optical purity was over 98% ee and the absolute configuration was L, as determined by HPLC analysis using a chiral column.<sup>9</sup>

Thus, the synthesis of L-[3-<sup>13</sup>C]alanine (**2**) from [<sup>13</sup>C]methyl iodide was accomplished by diastereoselective alkylation of Dellaria's oxazinone **1**. The D-form of **2** and other <sup>13</sup>C-isotopomers of L- or D-alanine should be similarly obtainable from the enantiomer of **1**, which could be derived from (*R*)-phenylglycinol, and <sup>13</sup>C-oxazinone, which we have previously prepared from sodium [1- or 2-<sup>13</sup>C]acetate.<sup>5</sup>

## Experimental

[<sup>13</sup>C]Methyl iodide (99 atom% <sup>13</sup>C) was supplied by Cambridge Isotope Laboratories. Melting points were measured on a Yanaco micro melting point apparatus and are uncorrected. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra were recorded on a JEOL GSX-400 Fourier-transform spectrometer. The chemical shifts were

reported in  $\delta$  values relative to tetramethylsilane (TMS) at 0 ppm in  $\text{CDCl}_3$  or sodium 3-trimethylpropionate- $d_4$  (TSP) at 0 ppm in  $\text{D}_2\text{O}$  on  $^1\text{H}$ -NMR and relative to  $\text{CDCl}_3$  at 77.0 ppm or TSP at 0 ppm in  $\text{D}_2\text{O}$  on  $^{13}\text{C}$ -NMR. IR spectra were recorded on a JASCO VALOR-III Fourier-transform spectrometer. EI- and HR-FAB-MS were obtained with a JEOL JMS-DX-302 double-focusing spectrometer and a Finnigan Autospec double-focusing spectrometer, respectively. HPLC analysis was carried out on a JASCO 800 Series HPLC system with a JASCO 875-UV detector. The column was Crown Pak CR(-) (150 mm  $\times$  4 mm i.d.), purchased from Daicel. HPLC analysis was performed at 20°C, and the other conditions were as described in a previous report.<sup>9</sup>

*(3S,5S)-2,3,5,6-Tetrahydro-3-[ $^{13}\text{C}$ ]methyl-5-phenyl-N-(tert-butyloxycarbonyl)-4H-1,4-oxazin-2-one (3)*

To a stirred solution of the oxazinone **1** (4.5 g, 16.2 mmol) in a 1:1 mixture of THF and DME (54 ml) was added dropwise a 1 M solution of NaHMDS in THF (27.6 ml, 27.6 mmol) over 30 min at  $-78^\circ\text{C}$ , and the mixture was stirred for 2 h. [ $^{13}\text{C}$ ]Methyl iodide was added dropwise to the resulting enolate solution. The mixture was stirred for 1 h at that temperature, then poured into saturated  $\text{NH}_4\text{Cl}$  with vigorous stirring. The aqueous layer was extracted with ether 3 times. The combined organic layers were washed with brine, dried over  $\text{Na}_2\text{SO}_4$  and evaporated. The residue was crystallized from hexane AcOEt to give **3** (4.74 g) in 92% yield. mp. 145.6–147.4°C;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ ,  $50^\circ\text{C}$ )  $\delta$ : 1.31 (brs, 9H), 1.63 (dd,  $J_{\text{C-H}} = 130.4$  Hz,  $J = 7.2$  Hz, 3H), 4.45 (d,  $J = 11.8$  Hz, 1H), 4.75 (dd,  $J = 3.1, 11.8$  Hz, 1H), 4.91 (m, 1H), 5.09 (m, 1H), 7.10 (m, 2H), 7.25–7.39 (m, 3H).  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ )  $\delta$ : 20.0. IR (KBr)  $\text{cm}^{-1}$ : 3030, 3000, 3965, 1754, 1680, 1495, 1398, 1370, 1355, 1297, 1160, 1120, 1068, 1039, 748, 739, 700. EI-MS (%)  $m/z$ : 292 ( $\text{M}^+$ , 9.3), 236 (18.1), 192 (16.9), 147 (14.8), 133 (12.7), 104 (68.4), 57 (100).

*L-[3- $^{13}\text{C}$ ]Alanine (2)*

A solution of the  $^{13}\text{C}$ -methyloxazinone **3** in 6 M HCl was heated at  $80^\circ\text{C}$  for 2 h. After lyophilization, the residue was dissolved in EtOH (40 ml), and 10% Pd-C (4.3 g, 4.1 mmol) was added. The mixture was shaken for 2 d under hydrogen (5 kg/cm<sup>2</sup>). The catalyst was removed by filtration through Celite, and the filtrate was evaporated. The residue was applied to a DOWEX 50W X8 ion-exchange column with EtOH. The column was washed with deionized water. The desired product was eluted with 1 M  $\text{NH}_4\text{OH}$ . The eluate was lyophilized, and the residue was crystallized from water-EtOH to give L-[3- $^{13}\text{C}$ ]alanine (**2**, 0.83 g) in 92% yield. mp. 226.2–227.7°C. Rt = 5.2 min (D-form gave Rt = 6.3 min, which was not detected).  $^1\text{H}$ -NMR ( $\text{D}_2\text{O}$ )  $\delta$ : 1.49

(dd, ( $J_{C-H} = 129.9$  Hz,  $J = 7.2$  Hz, 3H), 3.79 (dq,  $J_{C-H} = 4.6$  Hz,  $J = 7.2$  Hz, 1H).  $^{13}\text{C-NMR}$  ( $\text{D}_2\text{O}$ )  $\delta$ : 19.0. IR (KBr)  $\text{cm}^{-1}$ : 3082, 2602, 1623, 1591, 1455, 1411, 1359, 1306, 1233, 1153, 1110, 1012, 915, 847, 771, 648. HR-FAB-MS (glycerol): Calculated for  $\text{C}_2^{13}\text{C}_1\text{H}_8\text{NO}_2$ : 91.0586. found: 91.0592 ( $\text{MH}^+$ ).

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